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To cite this article: Akihiro Kojima *et al* 2008 *Meet. Abstr.* **MA2008-02** 27

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Novel Photoelectrochemical Cell with Mesoscopic Electrodes Sensitized by Lead-halide Compounds (11)

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In the research of the dye-sensitized solar cell (DSSC), narrow band gap semiconductors have been investigated as novel visible light sensitizer for oxide semiconductor electrodes. Principally, quantum dot sized chalcogenide materials (CdS, CdSe, etc) have been researched. We have previously reported that lead-halide perovskite compound, CH₃NH₃PbX₃ (X=Br, I), works as a visible light sensitizer for TiO₂ electrode in a photoelectrochemical cell, showing over 2% energy conversion efficiency ¹⁾. The stability of this system, however, is hardly adequate compared with the dye-sensitized system. One of the key techniques to enhance the stability of DSSC system is replacement of liquid type electrolyte with solid-state charge transport materials (p-type semiconductor, conductive polymer etc.). In this study, we attempted to fabricate a solid-state photovoltaic cell with polypyrrole-based conductive polymer material as charge transport layer for lead-halide compound sensitized photocell system.

Titania paste (Solaronix, SA) was spread on an F-doped SnO₂ (FTO) conductive glass by squeeze method and sintered at 500°C for 1h in the air. The photoelectrode (TiO₂/CH₃NH₃PbX₃) was prepared with the precursor solution consisted of CH₃NH₃X and PbX₂ on nano-porous TiO₂ electrode. Polypyrrole (undoped) composite with carbon black (PPCB), supplied from Sigma - Aldrich Inc., was employed as conductive polymer material. FTO glass was used as counter-electrode. A solid-state cell was fabricated to sandwich the PPCB material between TiO₂/CH₃NH₃PbX₃ and counter-electrode. Photovoltaic performance was measured under 100 mW cm⁻² irradiation using I-V measurement system (Peccell Technologies, Inc.). The effective area of solid-state cell was 0.238cm².

In the case of FTO/CH₃NH₃PbX₃-TiO₂/FTO system, photocurrent is hardly observed. On the other hand, photovoltaic characteristic was observed for the FTO/CH₃NH₃PbX₃-TiO₂/PPCB/FTO system. Fig.1 shows photocurrent-voltage characteristics of the solid-state cells and the photovoltaic performances are listed in Table.1. For the TiO₂/CH₃NH₃PbBr₃ system, the short-circuit photocurrent density (*J*_{sc}), open-circuit voltage (*V*_{oc}), fill factor (*FF*), and energy conversion efficiency (*η*) are 0.65mA cm⁻², 0.70V, 0.45, 0.21 %, and for the TiO₂/CH₃NH₃PbI₃ system, these are 1.39 mA cm⁻², 0.51 V, 0.52, and 0.37 %, respectively. The TiO₂/CH₃NH₃PbI₃ system showed higher photocurrent and energy conversion efficiency than TiO₂/CH₃NH₃PbBr₃ system. It is due to that CH₃NH₃PbI₃ have wide absorption region up to 800nm ²⁾. Comparing these results with those

employing iodine/iodide liquid electrolyte, photovoltaic performance of the solid-state systems is lower than that of the liquid electrolyte system. It is considered that conductive polymer material was not sufficiently incorporated into the depth of TiO₂ meso-porous electrode in this system. Despite low conversion efficiency of the present stage of study, this is a rare example of solid-state DSSC driven by inorganic sensitizer, which exhibits a fairly high photocurrent and voltage. To improve the photovoltaic characteristics, other conductive polymer materials will be examined together with developing a suitable pore-filling method for porous electrode.

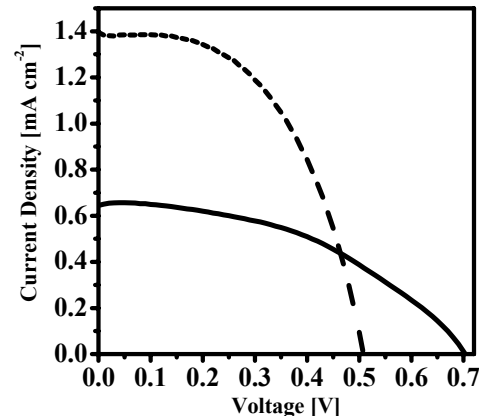


Fig.1 Current-voltage characteristics of polypyrrole based solid-state photocells using CH₃NH₃PbBr₃ (line) and CH₃NH₃PbI₃ (dot-line) as sensitizer for TiO₂ electrode .

Table.1 Photovoltaic performance of CH₃NH₃PbX₃ sensitized solid-state photocell with polypyrrole based conductive polymer as charge transport layer

<i>sensitizer</i>	<i>J</i> _{sc} (mA / cm ²)	<i>V</i> _{oc} (V)	<i>FF</i>	<i>Eff.</i> (%)
CH ₃ NH ₃ PbBr ₃	0.65	0.70	0.45	0.21
CH ₃ NH ₃ PbI ₃	1.39	0.51	0.52	0.37

- 1). A. Kojima, K. Teshima, Y. Shirai and T. Miyasaka, 210th ECS Meeting, Cancun, 2006.
2). A. Kojima, K. Teshima, Y. Shirai and T. Miyasaka, 212th ECS Meeting, Washington, 2007.