PAPER • OPEN ACCESS

 $Bi_{12}TiO_{20}$ crystallization in a Bi_2O_3 -Ti O_2 -Si O_2 -Nd₂O₃ system

To cite this article: S Slavov and Z Jiao 2018 J. Phys.: Conf. Ser. 992 012040

View the article online for updates and enhancements.

You may also like

- Raman spectroscopy of bismuth silicon oxide single crystals grown by the Czochralski technique Z Lazarevi, S Kosti, V Radojevi et al.
- <u>Mechanical Exfoliation of Plate-stratiform</u> <u>Structured Bi₄₂O₄₇Cl₂ for Enhanced</u> <u>Photocatalytic Performance</u> Mei Zhao, Zhipeng Guo, Congqi Fu et al.
- Effect of ambient argon pressure on the structural, optical and electrical properties of non-crystalline Se₈₅Te₃Bi₁₂ nano-thin films

Aditya Srivastava, Zishan H Khan and Shamshad A Khan





DISCOVER how sustainability intersects with electrochemistry & solid state science research



This content was downloaded from IP address 3.17.128.129 on 08/05/2024 at 20:17

Bi₁₂TiO₂₀ crystallization in a Bi₂O₃-TiO₂-SiO₂-Nd₂O₃ system

S Slavov^{1,3} and Z Jiao²

¹Department of Physics, University of Chemical Technology and Metallurgy, 8 Kl. Ohridski Blvd., 1756 Sofia, Bulgaria ²School of Environmental and Chemical Engineering, Shanghai University, Huanhua Building, 333 Nanchen Road, 200444 Shanghai, P.R. China

E-mail: stanislavslavov@hotmail.com

Abstract. Polycrystalline mono-phase bismuth titanate was produced by free cooling from melts heated to 1170 °C. The control over the initial amounts in the starting compositions in the system Bi₂O₃/TiO₂/SiO₂/Nd₂O₃ and over the thermal gradient of the heat process resulted in the formation of specific structures and microstructures of monophase sillenite ceramics. The main phase Bi₁₂TiO₂₀ belongs to the amorphous network groups based on oxides of silicon, bismuth and titanium. In this work, we demonstrated a way to control the crystalline and amorphous phase formation in bulk poly-crystalline materials in the selected system.

1. Introduction

Bismuth-titanate-based materials became very popular in recent years with a number of important applications, such as capacitors, sensors [1], piezoelectric, electro-optical and pyroelectric materials, relaxers, FERAM and DRAM storage devices and semiconductor devices [2].

The most popular methods for preparation are solid-state reactions, co-preparation, molten salt synthesis, mechanochemical synthesis. Addition of suitable additives is one way to modify their properties by controlling the phase formation. Jose Pineda-Flores et al. [3] have studied the ways to modify the properties of bismuth-titanate-based materials by controlling the phase formation and adding suitable additives, as Pr, Nd, Gd. Murugan et al. [4] discussed the glass-phase content influence on the dielectric properties of novel low-permittivity, fine-grain, pore-free and nanostructured materials. In general, the properties of bulk ceramic materials are affected by the phases involved, the grains size, the amount of glass phase and the grain boundary effects [5]. In the present work, we explored the possibility to obtain monophase polycrystalline ceramics using a bismuthsilicate amorphous network as a boundary-matrix on order to control the properties of these new bismuth-titanate materials.

2. Experimental

The synthesis of the bulk materials of the Bi_2O_3 -TiO₂-SiO₂-Nd₂O₃ system began by homogenizing for 15 min the starting oxides (Bi₂O₃, TiO₂, SiO₂ and Nd₂O₃, Alfa Aesar 99.99%). The melting was done in aluminum crucibles in a KTM-GSL1700X SiC tube furnace at a temperature of 1170°C.

³ To whom any correspondence should be addressed.

Content from this work may be used under the terms of the Creative Commons Attribution 3.0 licence. Any further distribution of this work must maintain attribution to the author(s) and the title of the work, journal citation and DOI. Published under licence by IOP Publishing Ltd 1

20th International Summer School on Vacuum, Electron and Ion Technologies	IOP Publishing
IOP Conf. Series: Journal of Physics: Conf. Series 992 (2018) 012040 doi:1	0.1088/1742-6596/992/1/012040

This study was part of a larger experiment aimed at obtaining monophase polycrystalline samples by melting and controlled cooling to room temperature. In the selected system, it is possible to obtain the following phases: $Bi_4Ti_3O_{12}$, $Bi_2Ti_2O_7$, $Bi_2Si_2O_7$, $Bi_{12}TiO_{20}$, $Bi_{12}SiO_{20}$, depending on the starting composition and the melting temperature. The remaining starting oxides (table 1) were selected with the idea of obtaining only one crystalline phase. $Bi_{12}TiO_{20}$, in combination with a Bi/Si/O amorphous network. In order to control the temperature gradient of the free cooling, it was carried out in graphite crucibles.

Table 1. Starting phase composition and melting conditions of selected samples in the system Bi_2O_3 -TiO_2-SiO_2-Nd_2O_3.

Sample index	Initial oxide composition (Raw materials in mol %)				t, °C /	Phases obtained after melting, according to
	Bi ₂ O ₃	TiO ₂	SiO ₂	Nd_2O_3	- time, mm	X-Ray Diffraction
А	77	14	0	9	1170/15	Bi ₁₂ TiO ₂₀
В	74	14	3	9	1170/15	Bi ₁₂ TiO ₂₀
С	74	3	14	9	1170/15	Bi ₁₂ TiO ₂₀

The phase composition was determined by X-ray diffraction using a Ridacu D/MAX2500V + PC apparatus, CuKa radiation (1.5406 Å). The microstructure was observed by scanning electron microscopy (SEM Hitachi SU1510). The structure of the samples was examined by Thermo Nikolet – Avatar 370 FT-IR equipment.

3. Results and discussion

As seen in the X-ray patterns (figure 1) only one phase, $Bi_{12}TiO_{20}$. was present in the three samples chosen. The SEM images (figure 2) show that the samples examined had a similar dense microstructure,



(C)
Image: Constraint of C

Figure 1. XRD patterns of samples A, B and C.

Figure 2. SEM images of samples A, B and C.

IOP Publishing

without visible cracks or cavities. The formation of separate crystals of different size is also seen.

According to the IR spectra (figure 3), the structures of samples A and B are similar; the band at about 830 cm⁻¹ is associated with the symmetric vibrations of the Ti-O linkages; and that at 600 cm⁻ with asymmetric and deformation vibrations of the TiO₆ octahedron [6]. Betch and White [7] reported that the band of 850 cm⁻¹ is related to symmetric vibrations of Ti-O bonds in addition to Nd together with the TiO₆ octahedral. The bands around 560 cm⁻¹ and 820 cm⁻¹ characterize a complex combination of a MO₄ tetrahedral and a BiOn polyhedral [8], typical for the sillenite phase $Bi_{12}TiO_{20}$. On the other hand, the bands near 900-800 cm⁻¹ and 900-1000 cm⁻¹, and the band near 1050-1100 cm⁻¹ are typical for structures with four terminal oxygens SiO⁴⁻ [13]. Jagannath Roy et al. [14] reported that the appearance of antisymmetric stretching vibrations of Si-O-Al and Si-O-Si networks can be connected with the peaks around 832 cm⁻¹ and 1112 cm⁻¹.



Figure 3. IR spectra of samples A, B and C.

The formation of a multi-component amorphous matrix with Si-O-Si-linkages

(1034 cm⁻¹, 1098 cm⁻¹), BiO₆ (480 cm⁻¹), Si-O-Ti linkages (900 cm⁻¹, 1034 cm⁻¹) and of depolymerized SiO_4 groups (890 cm⁻¹, 920 cm⁻¹) is due to the increased amount of SiO_2 in the starting composition of sample C. Thus, the different crystal sizes observed in he SEM images could be due to different SiO₂ content. In this sense, the results obtained by us directly correspond to the literature data, namely, that besides the formation of the sillenite bismuth-titanate phase Bi₁₂TiO₂₀, bismuth-silicate and titaniumsilicate multi-component amorphous structures are formed.

4. Conclusions

In this work we have investigated

1. The possibility to obtain monophase Bi₁₂TiO₂₀ polycrystalline ceramics,

2. The type and composition of the bismuth-silicate amorphous network as boundary-matrix between the crystalline phases.

This investigation indicates a way to create new monophase polycrystalline materials of the system Bi₂O₃-TiO₂-SiO₂-Nd₂O₃ using melting and controlled cooling to room temperature.

Acknowledgement

This work was supported by the Swap and Transfer, Erasmus Mundus Action 2 Mobility Lot 12, Grant ID number SAT 2542

IOP Conf. Series: Journal of Physics: Conf. Series **992** (2018) 012040 doi:10.1088/1742-6596/992/1/012040

References

- [1] Sedlar M and Sayer M 1996 Structural and electrical properties of ferroelectric bismuth titanate thin films prepared by the sol-gel method *Ceram. Inter.* **22** 241
- [2] Buhay H, Sinharoy S, Kasner W, Francombe M, Lampe D and Stepke E 1991 Pulsed laser deposition and ferroelectric characterization of bismuth titanate films *Appl. Phys. Lett.* 58 1470
- [3] Pineda-Flores J, Chavira E, Reyes-Gasga J, Gonzalezc A and Huanosta-Tera A 2003 Synthesis and dielectric characteristics of the layered structure Bi_{4-x}R_xTi₃O₁₂ (R_x=Pr, Nd, Gd, Dy) *J. Eu. Ceram. Soc.* 23 839–50
- [4] Murugan G, Subbanna G and Varma K 1999 Nanocrystallization of ferroelectrics bismuth unstated in lithium borate glass matrix *Mater. Sci. Lett.* **18** 1687-90
- [5] Huanosta A, Alvarez-Fregoso O, Amano Mexico E, Tabares-Muñoz C, Mendoza-Alvarez M and Mendoza-Alvarez J 1991 AC impedance analysis on crystalline layered and polycrystalline bismuth titanate J. Appl. Phys. 69 404
- [6] Du Y, Zhang M, Chen Q and Yin Z 2003 Investigation of size-driven phase transition in bismuth titanate nanocrystals by Raman spectroscopy *Appl. Phys.* A **76** 1099
- [7] Betch D and Write W 1997 Vibrational spectra of bismuth oxide and the sillenite-structure bismuth oxide derivative *Spectrosc. Acta* A **34** 505
- [8] Radaev S and Simonov V 1992 Structure of sillenite and atomic mechanisms of isomorphic substitution in them *Crystalograpy* **37** 4
- [9] Valant M and Suvorov D 2004 A stoichiometric model for sillenites Chem. Mater. 14 3471-6
- [10] Valant M, Medeu A and Suvorov D 2004 Isomorphic A-site substitution on sillenite-type compounds J. Am. Ceram. Soc. 84/4 677
- [11] Efendiev Sh, Kulieva T, Zomonov V, Chikgov M, Grandolfe M and Vecchia P 1981 Crystal structure of bismuth titanium oxide Bi₁₂TiO₂₀ *Phys. Stat. Soc.* A **74**
- [12] Abrahams S, Jamiesom P and Bernstain J 1967 Crystal structure of piezoelectric bismuth germanium oxide Bi₁₂GeO₂₀ J. Chem. Phys. 47/10 4034
- [13] Lazraev A 1972 Vibrational spectra and structure of silicates (Consultant Bureau, New York) pp 123-216
- [14] Roy J, Bandyopadhyay N, Das S and Maitra S 2011 Studies on the formation of mullite from diphasic Al₂O₃-SiO₂ gel by Fourier transform infrared spectroscopy *Iran J. Chem. Eng.* **30** 1