### **PAPER • OPEN ACCESS**

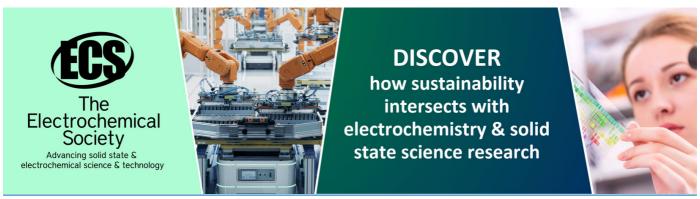
# Theoretical Study of Ultrafast Electron Dynamics in Amino Acids

To cite this article: D Ayuso et al 2015 J. Phys.: Conf. Ser. 635 112129

View the <u>article online</u> for updates and enhancements.

## You may also like

- Detection Limits for Chiral Amino Acids Using a Polarization Camera
   Claire Cook, Shane Byrne, Christian Drouet d'Aubigny et al.
- Charge migration induced by attosecond pulses in bio-relevant molecules Francesca Calegari, Andrea Trabattoni, Alicia Palacios et al.
- Biodistribution of phenylalanine labeled with gallium-68
   V K Tishchenko, V M Petriev and E D Stepchenkova



# Theoretical Study of Ultrafast Electron Dynamics in Amino Acids

## D. Ayuso\*1, A. Palacios\*, P. Decleva†, F. Martín\*‡2

\* Departamento de Química, Módulo 13, Universidad Autónoma de Madrid, Cantoblanco 28049, Madrid, Spain

† Dipartimento di Scienze Chimiche e Farmaceutiche, Università di Trieste and CNR-Istituto Officina dei

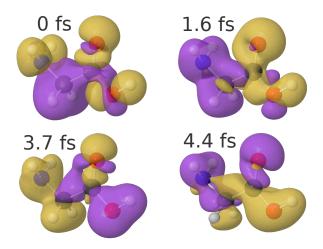
Materiali, 34127 Trieste, Italy

<sup>‡</sup> Instituto Madrileño de Estudios Avanzados en Nanociencia, Cantoblanco 28049 Madrid, Spain **Synopsis** Prompt ionization of large biological molecules may induce ultrafast charge migration along the molecular skeleton, preceding any nuclear rearrangement[1, 2]. This phenomenon has been recently observed in the amino acid phenylalanine [3] in a two-color pump probe experiment, where the production of ionic fragments was measured as a function of the time delay between the two pulses and charge fluctuations manifested as sub-4.5 fs oscillations in the quantum yield of a specific doubly charged fragment. We present our latest results in glycine, and compare with previous findings in phenylalanine [3]. We seek to perform a systematic study including larger aminoacids such as tryptophan.

The development of attosecond technology has enabled the real-time observation of electron motion in atoms, molecules and condensed matter [4]. In the case of large biological systems, such as amino acids, sudden ionization from a molecular orbital may trigger an ultrafast electronic response which is solely driven by electron correlation, as predicted in various theoretical works [1, 2]. However, due to the large bandwidth of attosecond pulses employed in state-ofthe-art experiments [3, 4], electrons are usually removed not from one but from many orbitals, thus leading to a coherent superposition of ionic states. Therefore, an accurate description of the ionization process becomes crucial in the description of these experimental observations.

We have performed a theoretical study of the field-free evolution of the electronic wave packet generated in amino acids (glycine, phenylalanine and tryptophan) upon the interaction with with attosecond pulses. The ionization amplitudes have been quantitatively determined for all open channels using the static-exchange DFT method [5] developed by Decleva and collaborators, which makes use of the Kohn-Sham DFT to describe molecular bound states and of the Galerkin approach to evaluate continuum wave functions in the field of the corresponding Kohn-Sham density. The evolution of the electronic wave packet has then been described by using a standard time-dependent density matrix formalism. The results of the numerical simulations clearly show the production of an ultrafast electronic response occurring in a sub-femtosecond time scale and are in good agreement with recent experimental observations [3]. We notice that, in contrast with most previous works, most valence and inner-valence electrons are efficiently ionized

and therefore the observed dynamics is that of a delocalized hole. We have further analyzed the role of the emitted electron in the hole migration mechanism and observed that even faster charge fluctuations (faster than the experimental resolution) might arise as a consequence of the interplay between the ground and continuum states.



**Figure 1.** Relative variation of the hole density generated by an attosecond XUV pulse in the amino acid glycine with respect to its time-averaged value (yellow:  $10^{-4}$ a.u., purple:  $-10^{-4}$ a.u.). Time is referred to the end of the pulse.

### References

- L. S. Cederbaum et al 1999 Chem. Phys. Lett. 307, 205-210.
- [2] F. Remacle et al 2006 P.N.A.S. 103, 6793-6798.
- [3] F. Calegari et al 2014 Science 346, 6207 336-339.
- [4] F. Krausz et al 2009 Rev. Mod. Phys. 81, 163.
- [5] M. Stener et al 2002 Chem. Phys. Lett. 351 469.

<sup>1</sup>E-mail: david.ayuso@uam.es <sup>2</sup>E-mail: fernando.martin@uam.es