OPEN ACCESS

Electron scattering from pyrimidine

To cite this article: Rafael Colmenares et al 2014 J. Phys.: Conf. Ser. 488 052022

View the article online for updates and enhancements.

You may also like

- Fluorescence emission from CsI(TI) crystals induced by high-energy heavy ions
- Yongtao Zhao, Yuanbo Sun, Yu Lei et al.
- <u>Total electron scattering cross sections for</u> pyrimidine and pyrazine as measured using a magnetically confined experimental system M C Fuss, A G Sanz, F Blanco et al.
- Methods for the synthesis of aza(deaza)xanthines as a basis of biologically active compounds
 D A Babkov, A N Geisman, A L Khandazhinskaya et al.





DISCOVER how sustainability intersects with electrochemistry & solid state science research



This content was downloaded from IP address 18.119.102.66 on 21/05/2024 at 07:39

Electron scattering from pyrimidine

Rafael Colmenares*[†], Martina C. Fuss[†], Juan C. Oller[§], Antonio Muñoz[§], Francisco Blanco[#], Diogo Almeida[‡], Paulo Limão-Vieira[‡], and Gustavo García[†]

^{*} Servicio de Oncología Radioterápica, Sección Radiofísica, Hospital Universitario Ramón y Cajal, 28034 Madrid, Spain

[†] Instituto de Física Fundamental, Consejo Superior de Investigaciones Científicas, 28006 Madrid, Spain

⁸ Centro de Investigaciones Energéticas, Medioambientales y Tecnológicas (CIEMAT), 28040 Madrid, Spain

[#] Departamento de Física Atómica, Molecular y Nuclear, Universidad Complutense de Madrid, 28040 Madrid [‡] Atomic and Molecular Collisions Laboratory, Department of Physics, New University of Lisbon, 2829-516 Caparica,

Portugal

Synopsis Electron scattering from pyrimidine $(C_4H_4N_2)$ was investigated over a wide range of energies. Following different experimental and theoretical approaches, total, elastic and ionization cross sections as well as electron energy loss distributions were obtained.

Being the molecular base of biomolecules like cytosine, thymine, uracil, thiamin, and alloxan, pyrimidine is an interesting candidate for studying its electron interaction properties.

Here, we present experimental and theoretical results on electron scattering from gaseous pyrimidine in different energy ranges. First, total cross sections (CS) were determined by measuring the beam attenuation in the energy range 10-500 eV. This was done using a newlyconstructed apparatus confining the electron beam magnetically along the central axis before, during and after scattering.

Electron-impact ionization experiments were carried out combining simultaneous electron and ion measurements with a time of flight analysis of the ionic fragments produced.

With an optical potential method based on the independent atom model and including the screening-corrected additivity rule [1], total and integral elastic and inelastic CS (1 eV - 10 keV) were also calculated in order to complement the experimental data. Generally, good agreement with our experimental results is observed in the common energy range. Furthermore, IAM-SCAR elastic differential CS are presented.

Electron Energy Loss (EEL) spectra were measured in the range 30-2000 eV using a transmission beam technique that measures a mixture of small angles [2]. In fig. 1, some examples of the energy loss distributions obtained are displayed. The average energy loss calculated from these spectra, together with our calculated inelastic cross section, has been used for

computing the electron stopping power of pyrimidine for energies of 20-3000 eV.

Finally, the present theoretical and experimental results, complemented by data available in the literature where appropriate, will be used as the input data for the Low-Energy Particle Track Simulation (LEPTS), a Monte Carlo programme code capable of simulating electron and positron interactions in different media at the molecular level.



Figure 1. Electron energy loss distributions measured for pyrimidine at various incident energies.

References

- [1] F Blanco and G García 2003 Phys. Rev. A 67 022701; 2004 Phys. Lett. A 330 230-237
- [2] R Colmenares, A G Sanz, M C Fuss, F Blanco and G García 2013 Appl. Radiat. Isotopes http://dx.doi.org/10.1016/j.apradiso.2013.01.025

¹E-mail: g.garcia@iff.csic.es